More Mixed Ionic and Covalent Compound Nami

Name 'em:

1) Br_2 Bromine

2) $(NH_4)_2SO_4$ Ammonium sulfate

3) HNO_2 Nitrous Acid

4) $Pb(CO_3)_2$ Lead (IV) Carbonate

5) N_2O dinitrogen Monoxide

6) ZnF_2 Zinc Flouride

7) CCl_4 Carbon Tetrachloride

8) H_2S Hydro sulfic acid

9) $Al(OH)_3$ Aluminum Hydroxide

10) K_2O Potassium Oxide

Formulas, please:

11) sulfur trioxide SO_3

12) tin(II) sulfide SnS

13) magnesium acetate $Mg(C_2H_3O_2)_2$

14) hydrophosphoric acid H_3P

15) cadmium chloride $CdCl_2$

16) iron(III) phosphate $FePO_4$

17) dinitrogen trioxide N_2O_3

18) fluorine F_2

19) vanadium(V) bicarbonate $V(HCO_3)_5$

20) ammonia NH_3

Mixed Ionic and Covalent Naming Practice

e formulas provided, name each of the following compounds:

$_2O_3$ Tellurium trioxide

S_2 Carbon disulfide

$_2(OH)_3$ Iron (III) Hydroxide

$_2O$ Calcium Oxide

Se_3 Tetraphosphorous triselenide

S_3 Vanadium (III) Sulfide

$_2SO_4$ Copper (II) Sulfate

$H_4)_3PO_4$ Ammonium Phosphate

F_4 Cadmium (IV) Flouride

NO_2 Sodium Nitrite

formulas of the following covalent compounds using the names shown:

mine pentachloride $BrCl_5$

aphosphorus trisulfide T_4S_3

um acetate $LiC_2H_3O_2$

phosphide Zn_3P_2

um nitride Ba_3N_2

ntium chloride $SrCl_2$

enic pentafluoride AsF_5

inum hydroxide $Al(OH)_3$

alt (III) carbonate $Co_2(CO_3)_3$

ganese (II) bicarbonate $Mn(HCO_3)_2$